



Universiteit Leiden

7D Quantum Scattering Calculations for H₂@Cu(111)



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Leiden Institute of Chemistry

Thursday November 22th, 2012

Outline

- 1 Quantum Scattering
 - Scattering from a square potential barrier
- 2 The 7D H₂@Cu(111) Model
 - Dynamical model and Potential Energy Surface
- 3 7D Quantum Dynamical Results
 - Dissociative Chemisorption and Scattering Results
- 4 Open questions
 - Directions for Possible Improvements

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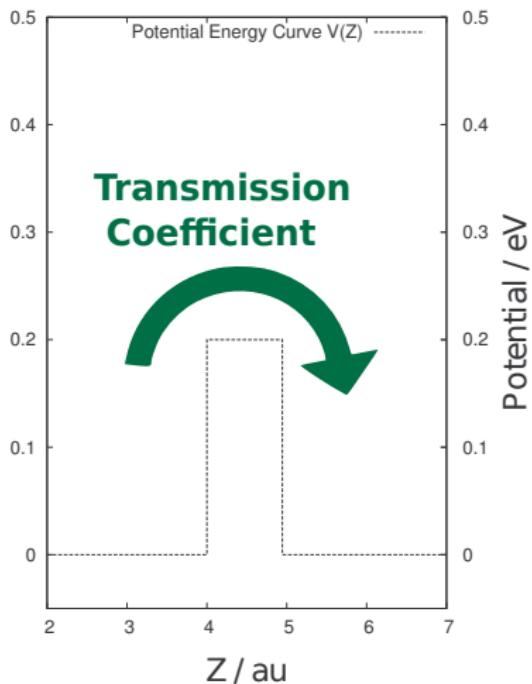
3 7D Quantum Dynamical Results

- Dissociative Chemisorption and Scattering Results

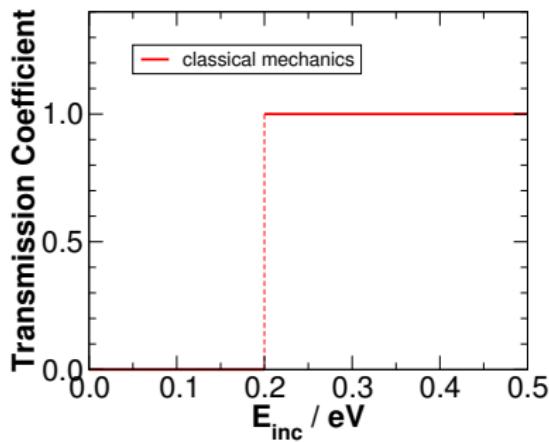
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The Square Potential Barrier



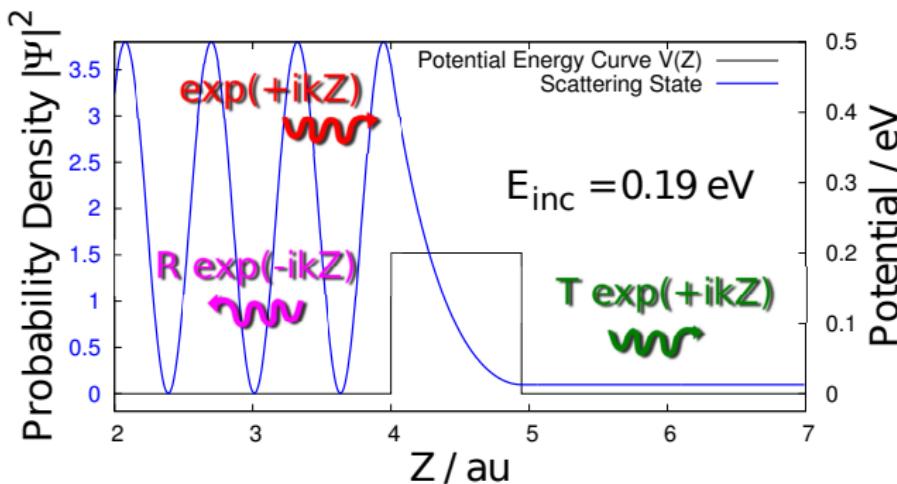
- **particle:** mass 1.0 AMU
given momentum p
- **square barrier:** $E_{\text{barrier}} = 0.2 \text{ eV}$
 $L = 0.5 \text{ \AA}$
- **Transmission Coefficient:**
probability to overcome the barrier



Time Independent Quantum Scattering

Scattering States $\psi^+(x, E)$

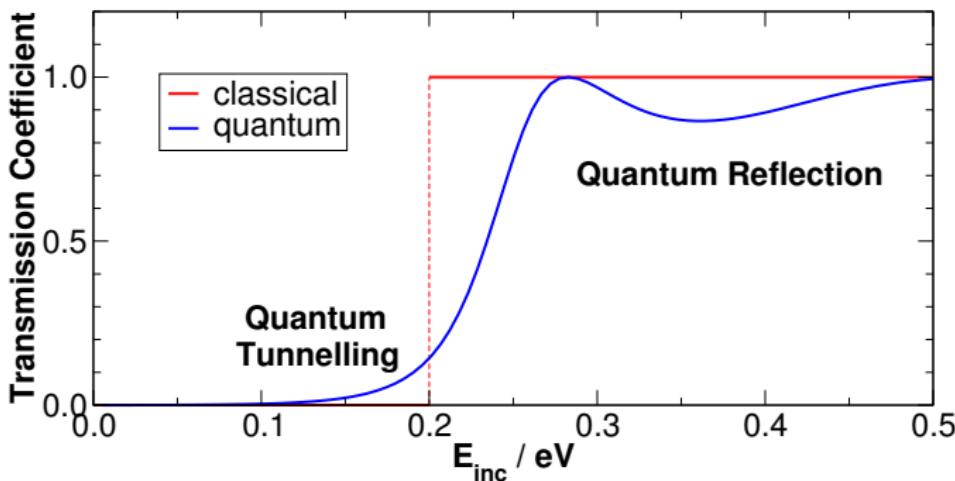
- **Time independent** solutions: $\mathcal{H}\Psi = E\Psi$
for $Z \rightarrow \pm\infty$, plane waves with fixed momentum $p = \hbar k$
- Superposition of waves: **incident**, **reflected** and **transmitted**
- Transmission = $|\text{transmitted}|^2 / |\text{incident}|^2$



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Time Dependent Quantum Scattering

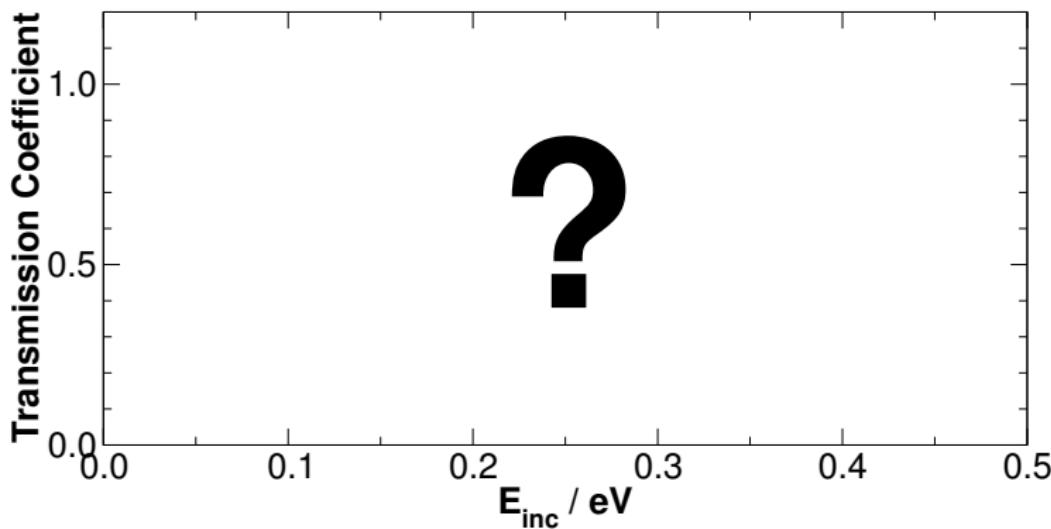
Time Dependent Wavepacket $\psi(x, t)$

- **Time dependent** solutions: $i\hbar \frac{\partial}{\partial t} \Psi = \mathcal{H} \Psi$
at t=0, gaussian wavepacket with **gaussian p distribution**
- Continous E_{inc} distribution: how to extract Trasmission vs E_{inc} ?

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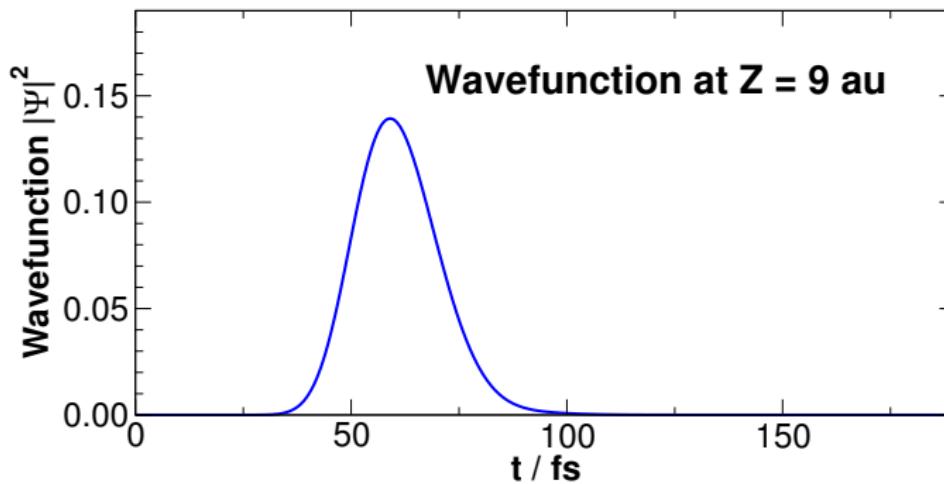


Time-Energy Mapping

Time-Energy Fourier Transform $\psi^+(x, E) \Leftrightarrow \psi(x, t)$

time-dependent wavepacket and scattering states are related by FT

$$\psi^+(\tilde{x}, E) = \frac{\sqrt{E}}{\pi\sqrt{2M}\phi(E)} \int_{-\infty}^{+\infty} dt e^{\frac{i}{\hbar}Et} \psi(\tilde{x}, t)$$

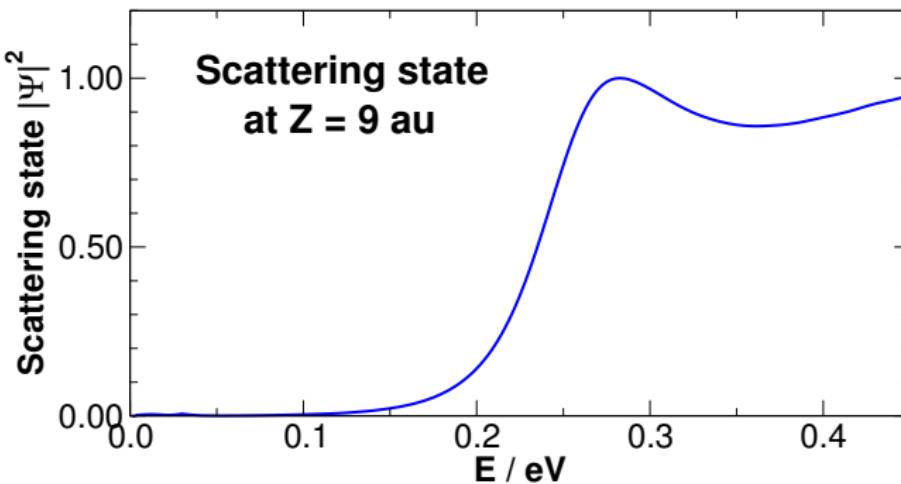


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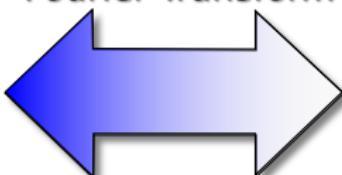


Quantum Scattering Methods

Time Independent

- Scattering states
- Time Indep SE:
 $\mathcal{H}\Psi = E\Psi$
- Asymptotic condition, $x \rightarrow \infty$

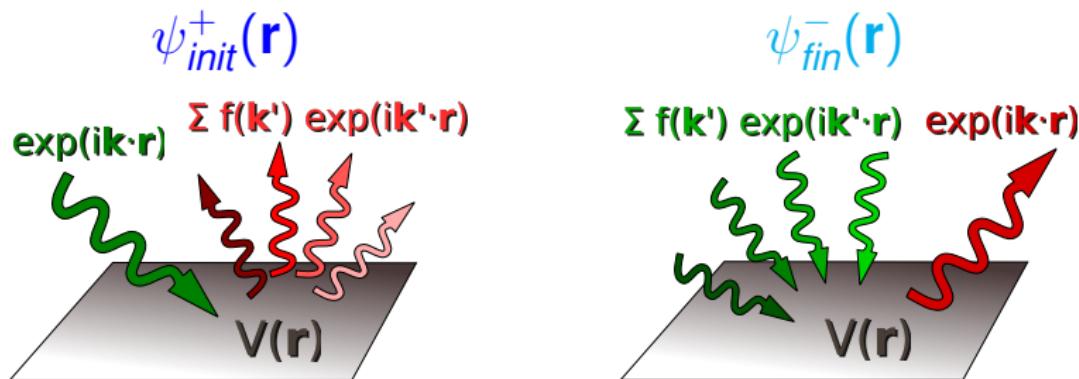
Time - Energy Fourier Transform



Time Dependent

- Wavepackets
- Time Dep SE:
 $i\hbar \frac{\partial}{\partial t} \Psi = \mathcal{H}\Psi$
- Initial condition,
 $\psi(t = 0) = \psi_0$

Scattering Matrix



Scattering Matrix $S(fin \leftarrow init)$

- the S-matrix is the **overlap of in and out** scattering states

$$S(fin \leftarrow init) = \int d\mathbf{r} \psi_{fin}^-(\mathbf{r})^* \psi_{init}^+(\mathbf{r}) = \langle \psi_{fin}^- | \psi_{init}^+ \rangle$$

- The **scattering probability** $P(fin \leftarrow init)$ is given by

$$P(fin \leftarrow init) = \|S(fin \leftarrow init)\|^2$$

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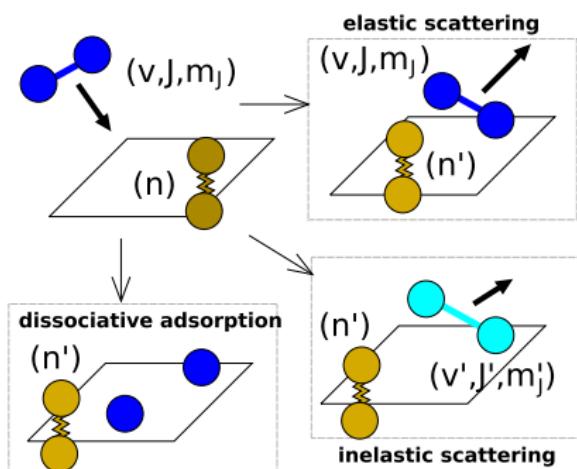
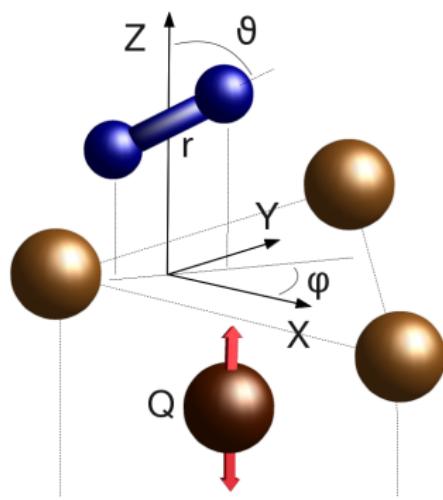
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The Dynamical Model

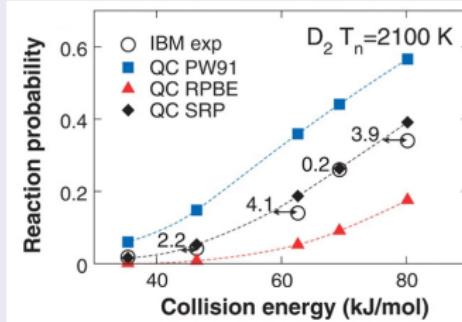
- 7 DoF model
 - ① 6 DoF of H₂: (r,Z,X,Y, ϑ , φ)
 - ② 1 Dof of Cu surface: 2nd layer, \perp direction
- Quantum Wavepacket Calculations



Potential Energy Surface

DFT method

- **2x2 Supercell**
- **Plane Waves:**
cutoff 350 eV, 8x8x1 \mathbf{k} points
- **Functional:** mixed PW91/RPBE
Specific Reaction Parameter



- Calculations with **DACAPO** code

PES

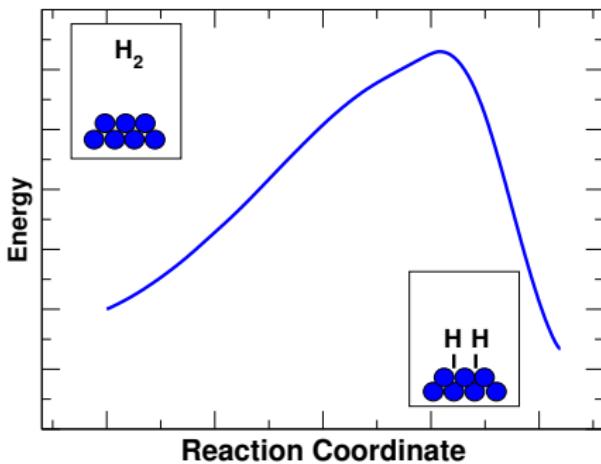
- **6D Potential** at fixed $Q = -0.2 \text{ \AA}$, 0.0 \AA , 0.2 \AA with **Corrugation Reducing Procedure**
- **Q interpolation** with parabola

^aC. Díaz *et al.* Phys. Chem. Chem. Phys. **12**, 6499 (2010)
^bH.F. Busnengo *et al.* J. Chem. Phys. **112**, 7641 (2000)

Effect of Lattice DoF on Activated Chemisorption

Effects of lattice displacement on reaction path

- ① electronic coupling: barrier height
- ② mechanical coupling: barrier position



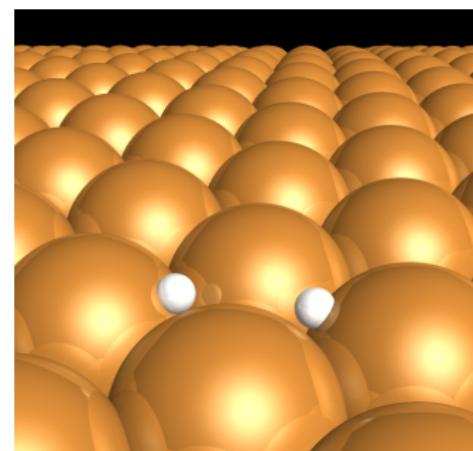
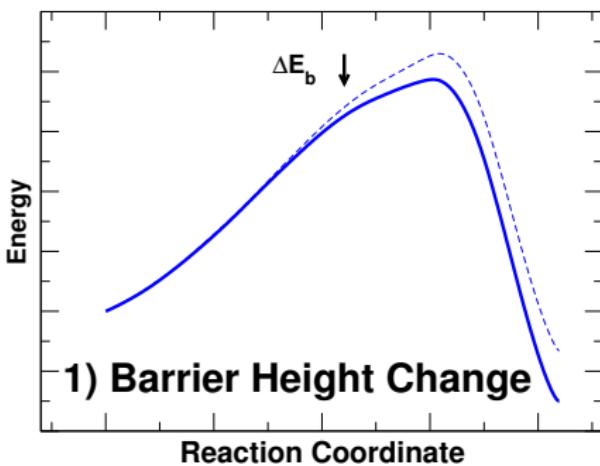
^aS. Nave and B. Jackson, J. Chem. Phys. **130**, 054701 (2009)

^bM.Bonfanti, C.Díaz, M.F.Somers and G.J. Kroes, PCCP, **13**, 4552 (2011)

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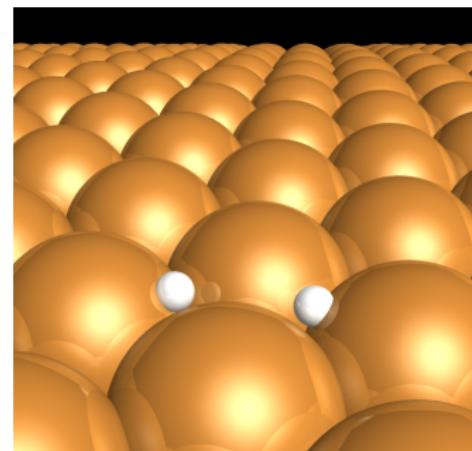
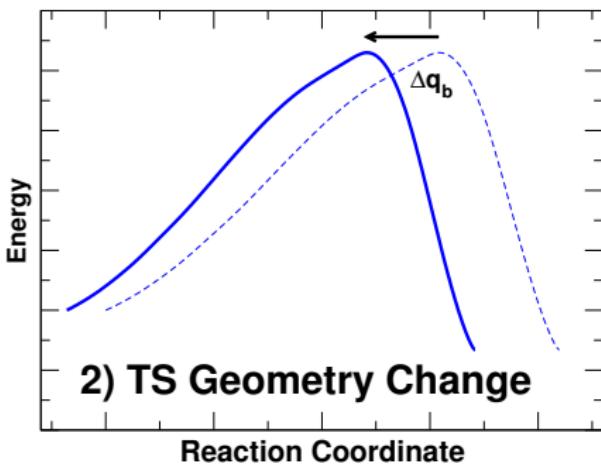
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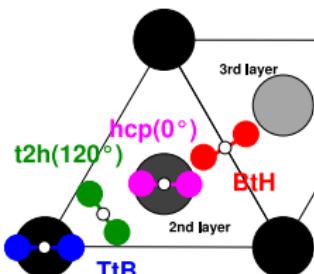


^aS. Nave and B. Jackson, J. Chem. Phys. **130**, 054701 (2009)

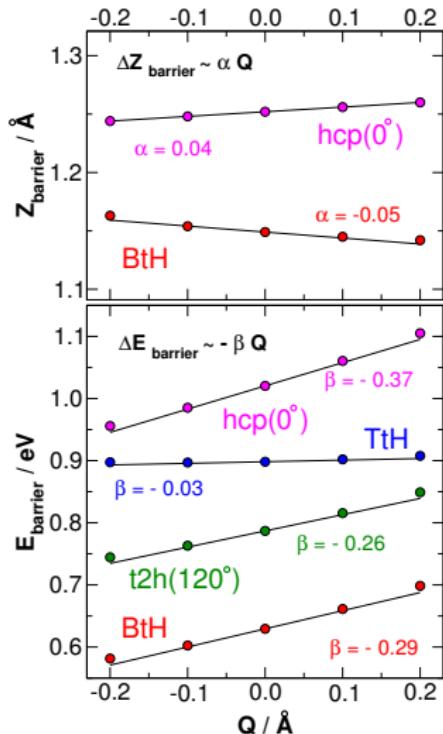
^bM.Bonfanti, C.Díaz, M.F.Somers and G.J. Kroes, PCCP, **13**, 4552 (2011)

Effect of 2nd Layer Cu \perp motion

Barrier	Q (\AA)	Geometry			ΔE (meV)
		Θ ($^\circ$)	Δr (\AA)	ΔZ (\AA)	
BtH	-0.2	90.	-0.03	0.01	-48.
	0.2	90.	0.03	-0.01	69.
t2h	-0.2	90.	-0.02	0.00	-42.
	0.2	90.	0.02	0.00	62.
TtH	-0.2	90.	-0.01	0.00	-1.
	0.2	90.	0.00	0.00	10.
hcp	-0.2	90.	-0.01	-0.01	-65.
	0.2	90.	0.00	0.01	85.



- Negligible change in barrier geometry
- Displacement mainly in bond length r



^a M.Bonfanti, C.Díaz, M.F.Somers and G.J. Kroes, PCCP, 13, 4552 (2011)

Vibrational Sudden Approximation

- VSA scattering matrix

$$S_{\text{sudden}}(\mathbf{f}, n' \leftarrow \mathbf{i}, n) = \int dQ \chi_{n'}^*(Q) S_{\text{frozen Q}}(\mathbf{f} \leftarrow \mathbf{i}, Q) \chi_n(Q)$$

- Summing over the final vibrational state n'

$$P_{\text{sudden}}(\mathbf{i}, n) = \int dQ |\chi_n(Q)|^2 P_{\text{frozen Q}}(\mathbf{i}, Q)$$

- We expect VSA to hold with **small coupling** and **slow vibrations**

From a practical point of view...

- **Compute 6D S-matrix** on a set of fixed Q (typically 20)
- Evaluate average with **Q quadrature**

^aJ.M. Bowman *et al.*, J. Chem. Phys. 71, 2270 (1979)

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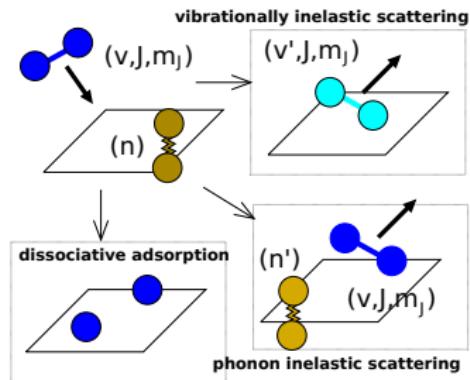
Results

Dynamical models

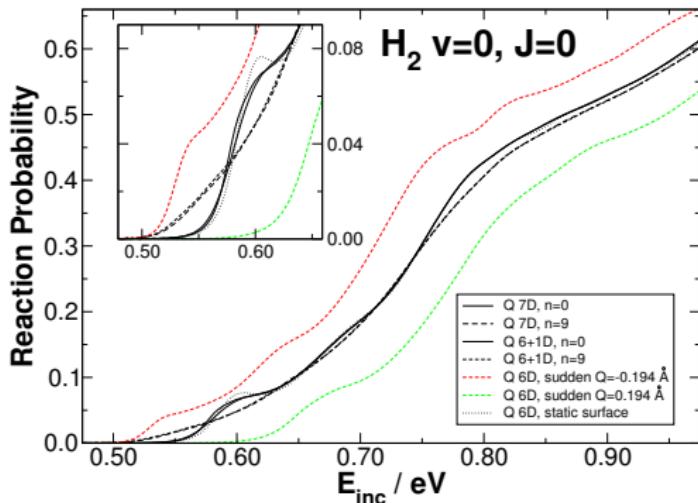
- **BOSS**: quantum dynamics with static surface,
Q frozen in equilibrium position
- **7D**: quantum dynamics with full 7D model,
Q motion explicitly included in the dynamics
- **VSA**: quantum dynamics with Vibrational Sudden Approximation for the
 7^{th} DoF

Observables

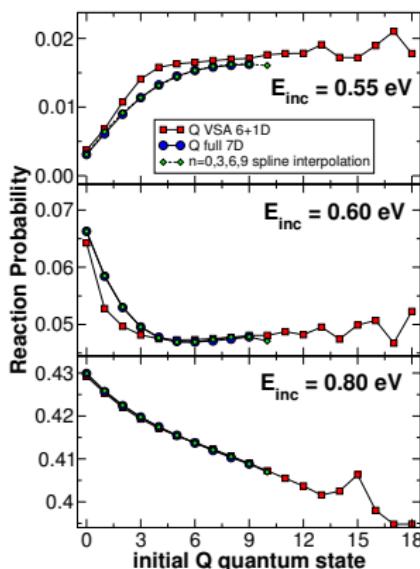
- Dissociative chemisorption
- Vibrationally Inelastic Scattering
- Phonon Inelastic Scattering



Dissociative Chemisorption: BOSS, 7D and VSA



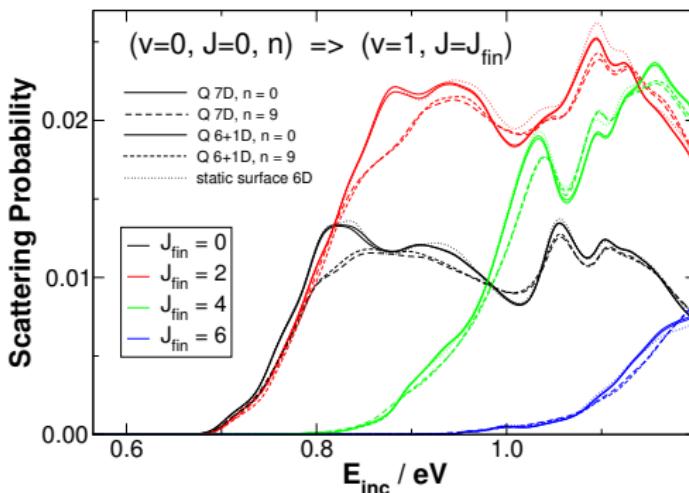
- H₂ v=0, J=0
- Static surface vs moving surface
- 7D and VSA



- Very small difference between static surface and Q ground state
- Broadening and smoothening with vibrational excitation
- VSA and 7D in very good agreement

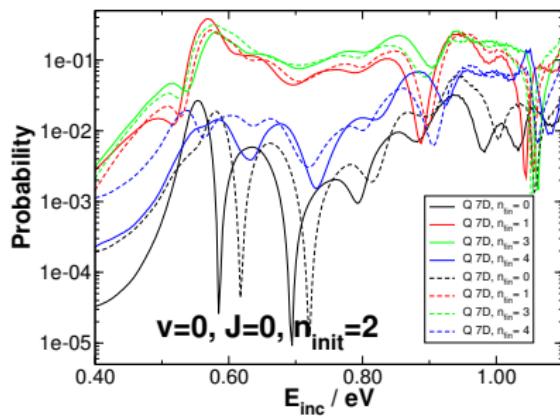
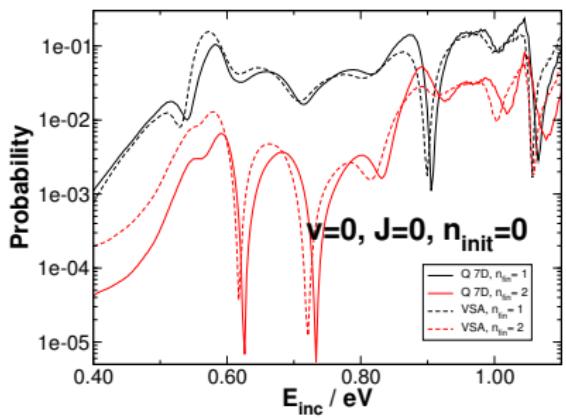
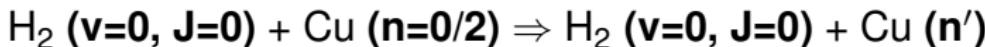
Vibrationally Inelastic Scattering: BOSS, 7D and VSA

$$\text{H}_2 \ (\nu=0, J=0) \Rightarrow \text{H}_2 \ (\nu=1, J=0/2/4/6)$$



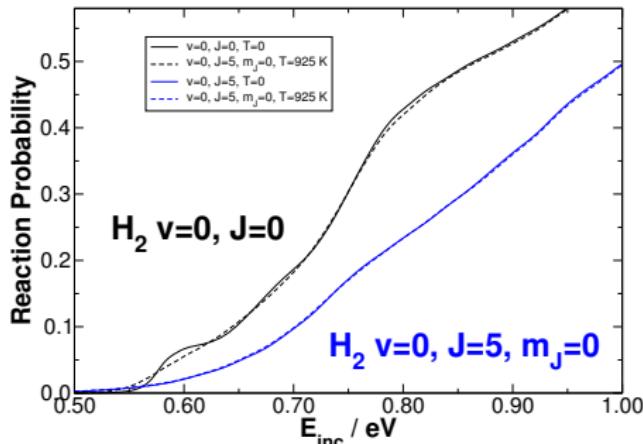
- Small difference between static surface and Q ground state
- Slight decrease of the probabilities with vibrational excitation
- VSA and 7D in **very good agreement**

Phonon Inelastic Scattering: 7D and VSA



- stricter test on VSA: molecule-surface energy transfer
- reasonable agreement, behaviour well reproduced but some quantitative mismatch

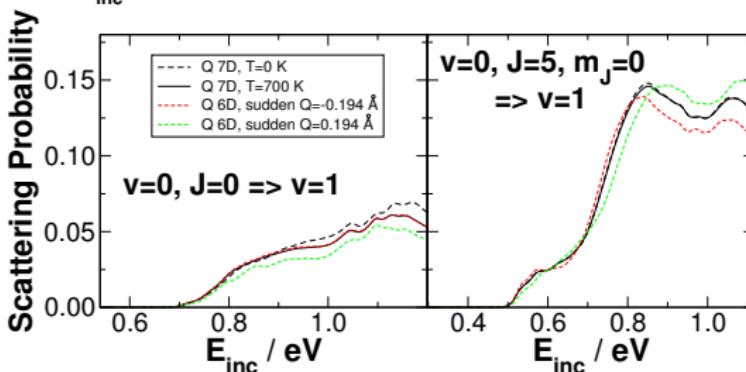
Thermal Averaged Results



Thermal distribution of initial n states:

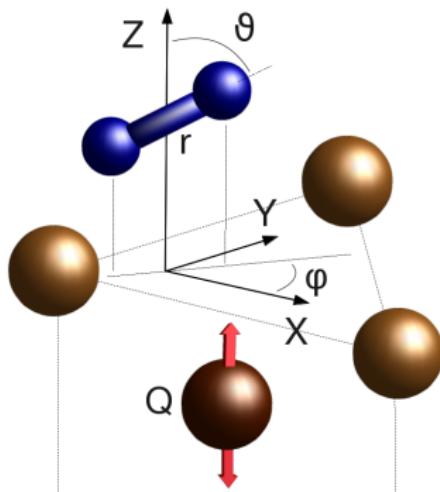
$$P = \sum_n \omega_n p_n \quad \omega_n = \frac{e^{-\beta E_n}}{Z}$$

- Very small effect on reaction probability
- Cannot explain the observed increase of inelastic scattering



Conclusions

- 1 VSA is a very promising approach
 - Very good agreement for reaction and scattering probability
 - Even qualitative description of phonon-inelastic scattering
- 2 Model is not sufficient to describe experimental observations
 - Very small broadening of reaction probability
 - No increase of vibrationally inelastic scattering



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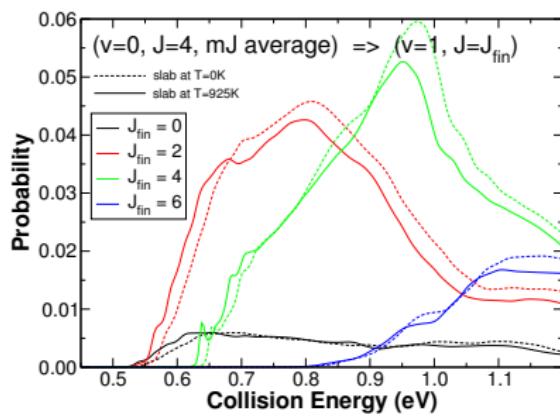
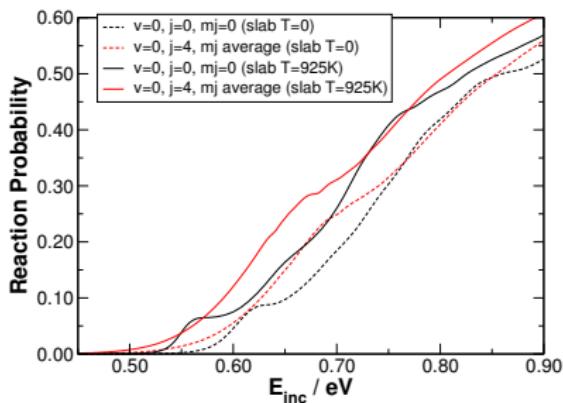
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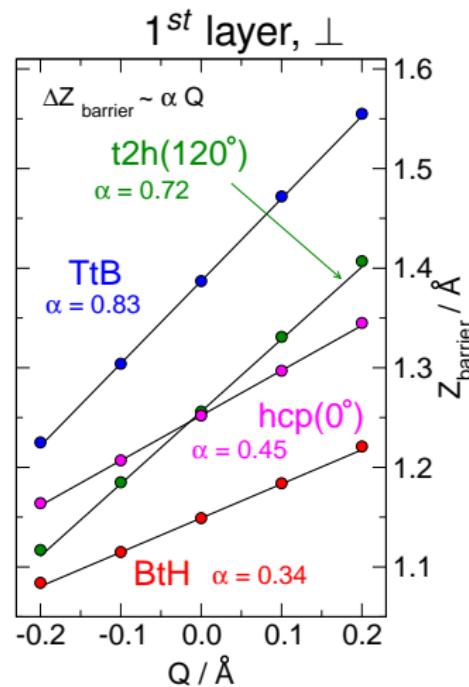
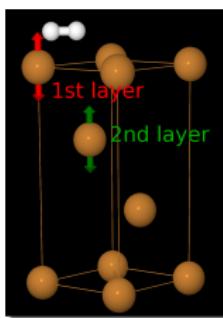
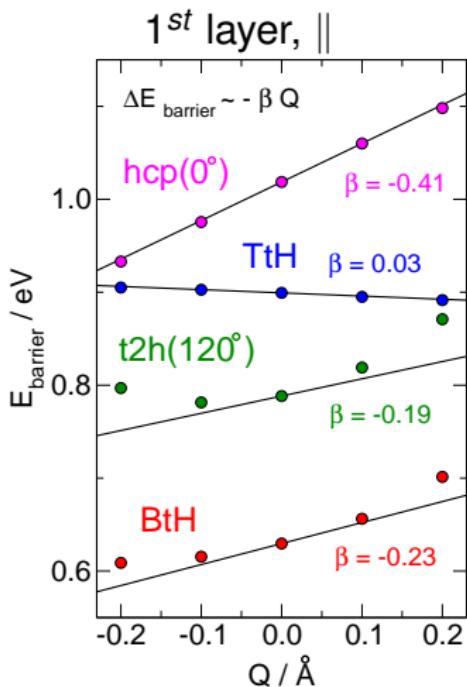
Thermal Expansion

Reaction and scattering probability with two 6D PES:

- Cu atoms in equilibrium position (0 K)
- Expanded unit cell according to the experimental thermal expansion at 925 K



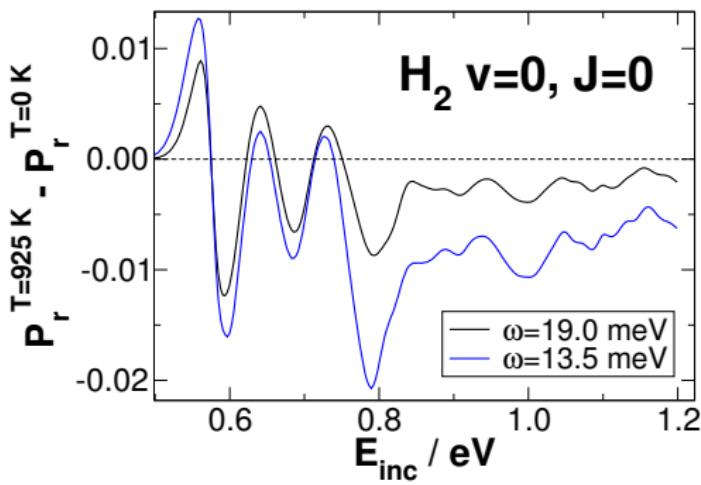
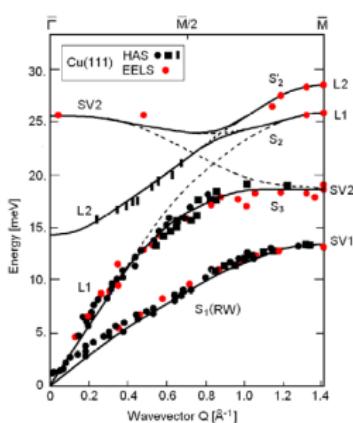
First layer atom motion



- Both mechanical (\perp) and electronic ($||$) effects

Vibrational Frequency of the DoF

- Ambiguous definition of ω in localized picture
- “Antisymmetric” surface phonon modes with lower ω (~ 13 meV)
- Simulation with $\omega=13$ meV instead of 19 meV



Acknowledgments

- Geert-Jan Kroes
- Mark Somers
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- Fabio Busnengo
- Mark Wijzenbroek
- Arobendo Mondal

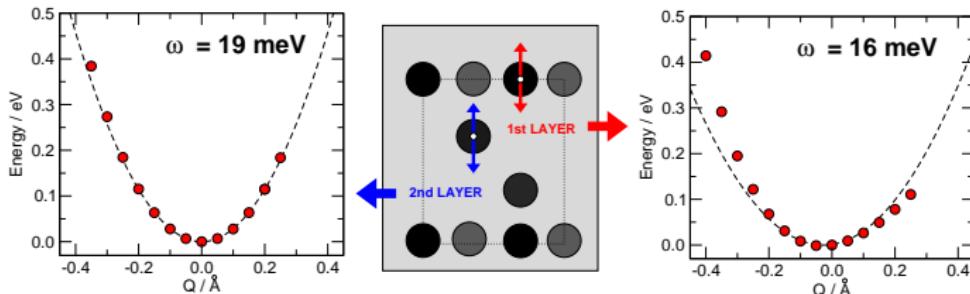
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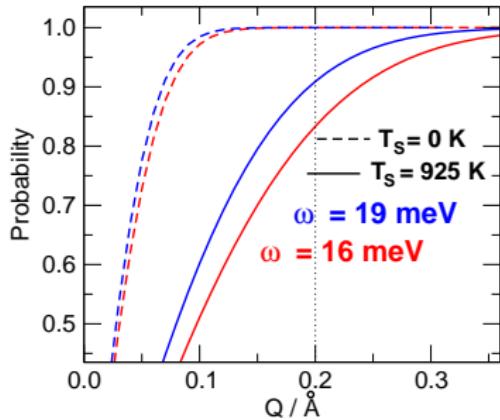
Vibrations of Cu atoms



Assuming an harmonic model
and Boltzmann's distribution:

$$P = \sum_n \omega_n p_n \quad \omega_n = \frac{e^{-\beta E_n}}{Z}$$

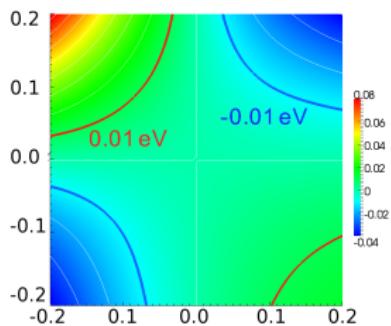
$$p_n = \int_{-x_0}^{x_0} |\psi_n|^2 dx$$



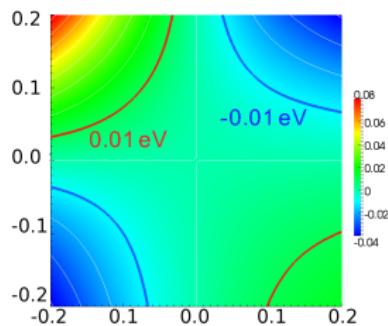
Surface DOF: Potential Energy Coupling

$$V_{2D}(Q_1, Q_2) = V_{1D}(Q_1) + V_{1D}(Q_2) + V_{coupling}(Q_1, Q_2)$$

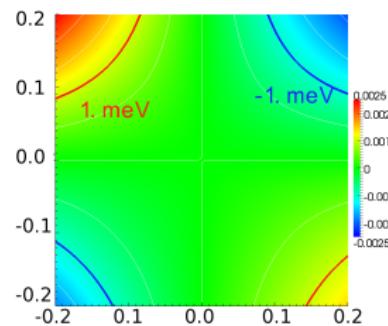
Contour plots of $V_{coupling}$



$Q_1 \rightarrow 1^{st}$ layer
 $Q_2 \rightarrow 2^{nd}$ layer



$Q_1 \rightarrow 2^{nd}$ layer
 $Q_2 \rightarrow 3^{rd}$ layer



$Q_1 \rightarrow 1^{st}$ layer
 $Q_2 \rightarrow 3^{rd}$ layer

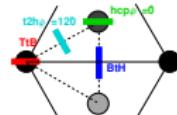
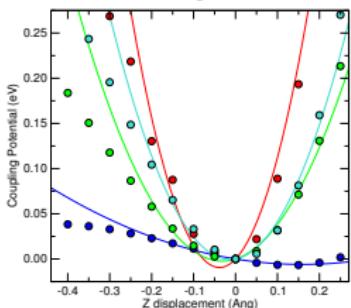
H₂-Surface Coupling Potential

$$V_{7D}(\xi, Q) = V_{H2@Cu}(\xi) + V_{strain}(Q) + V_{coup}(\xi, Q)$$

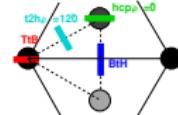
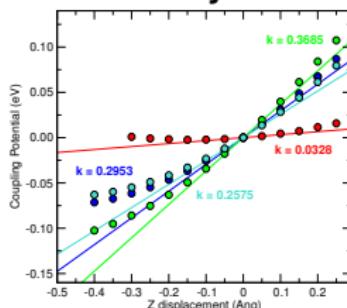
$$\xi = (X, Y, Z, r, \vartheta, \varphi)$$

compare different surface DOF by comparing
 $V_{coup}(\xi_{bar}, Q)$ vs Q $\xi_{bar} \rightarrow$ lowest barrier geometry

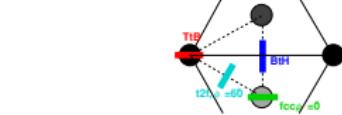
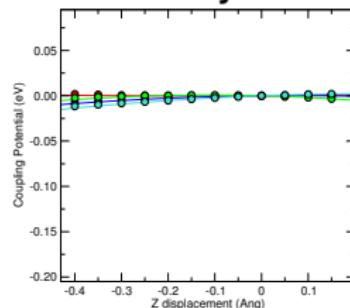
1st Layer Cu



2nd Layer Cu

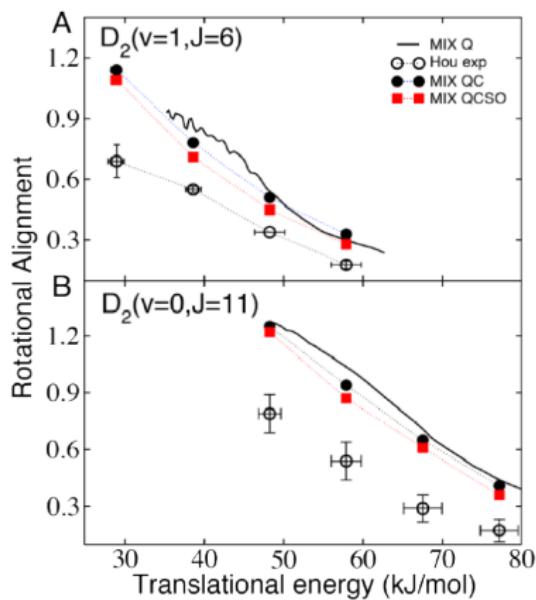


3rd Layer Cu



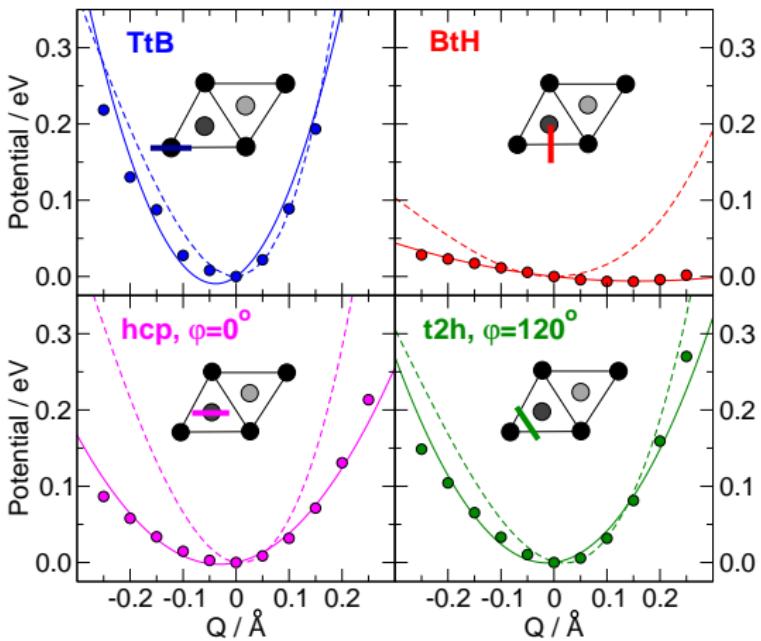
Lattice Motion: Surface Oscillator Model

$$V_{SO}(X, Y, Z, r, \vartheta, \varphi, Q) = V_{6D}(X, Y, Z - Q, r, \vartheta, \varphi) + \frac{1}{2} \mu \omega^2 Q^2$$



- V_{6D} is shifted along Z , as if the whole surface move with Q
- the potential for the surface vibration is quadratic

1st Layer vs SO Model

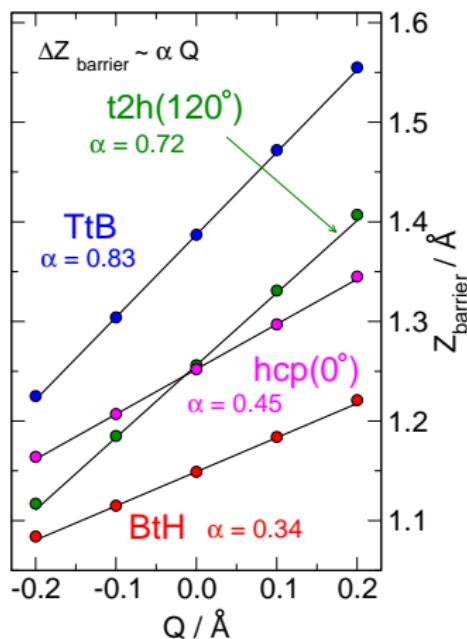


Agreement between SO Model and 1st layer atom motion

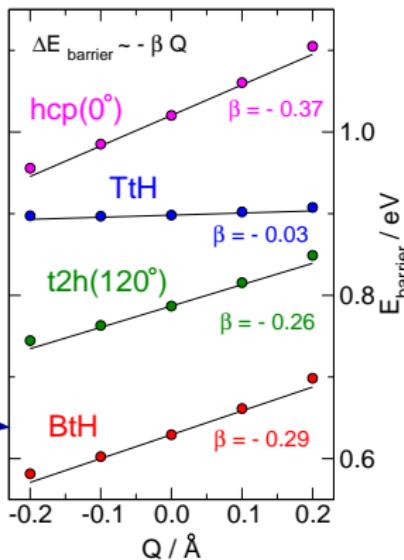
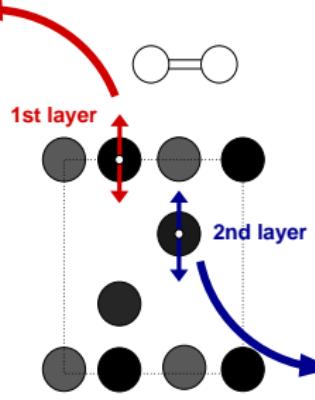
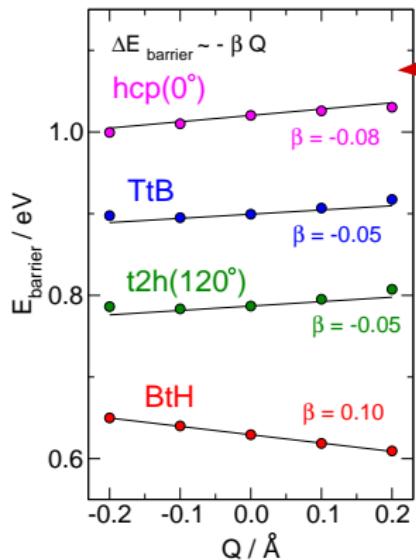
Barrier Displacement - 1st Layer Cu motion

Barrier	Q (Å)	Geometry			ΔE (meV)
		θ (°)	Δr (Å)	ΔZ (Å)	
BtH	-0.2	90.	0.01	-0.06	21.
	0.2	90.	0.01	0.07	-20.
t2h	-0.2	90.	-0.04	-0.14	-1.
	0.2	90.	0.06	0.15	21.
TtB	-0.2	90.	-0.04	-0.16	-2.
	0.2	90.	0.06	0.17	18.
hcp	-0.2	83.	-0.04	-0.09	-21.
	0.2	96.	0.06	0.09	10.

- Bigger ΔZ than Δr
- ΔZ similar to Q near the TOP Cu atom
- Tilting angle for hcp barrier, mostly **geometrical effect** $\Delta\theta_{geom} = 4.4^\circ$

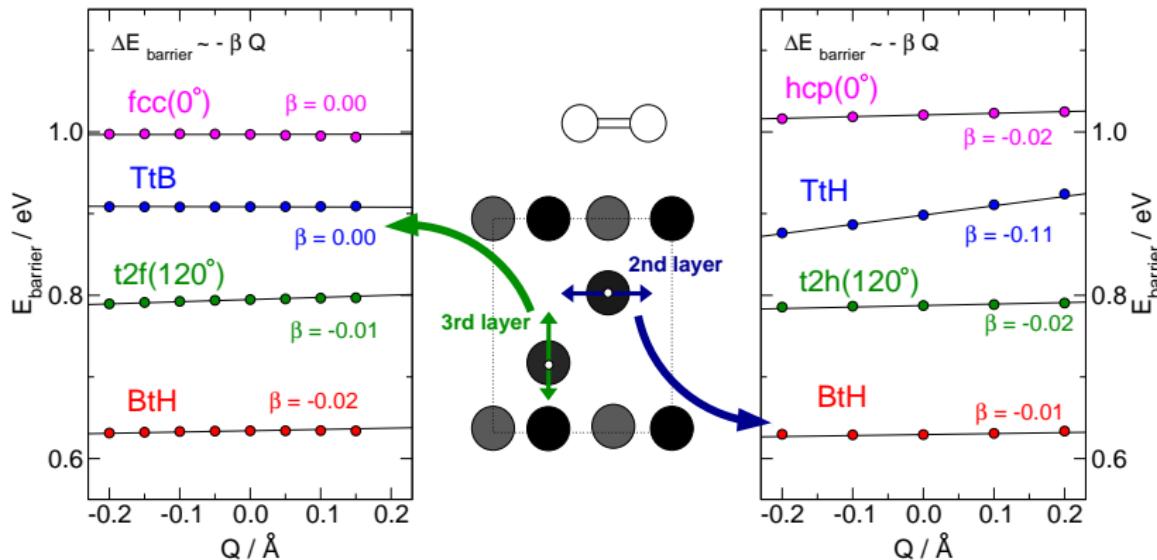


Barrier Height - 1st Layer vs 2nd Layer Cu motion



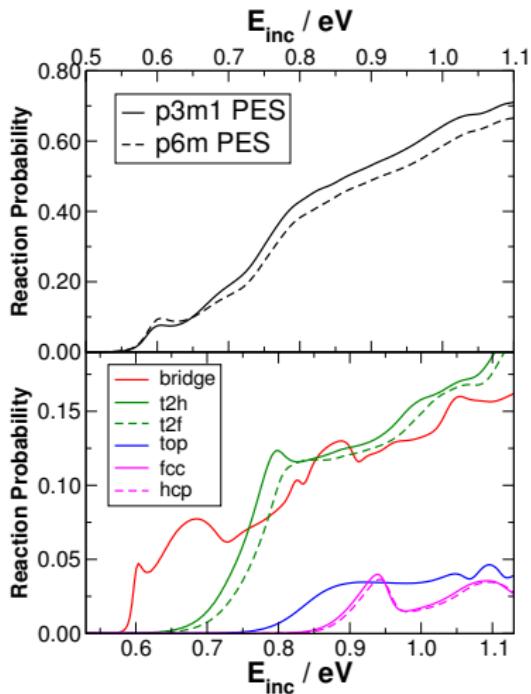
E_{barrier} more sensitive to 2nd Layer motion!

Other lattice coordinates



E_{barrier} mostly independent of other lattice coordinates

Reaction Probability: 6D and 4D calculations



- ➊ **Static Surface** P_{reaction} , comparison between **p6m** (previous PES with hcp = fcc) and **p3m1** (current PES)
- ➋ 4D calculations with **X, Y fixed** in high symmetry sites, weighted and with Dai-Light ZPE correction

- **p6m / p3m1 difference** is consistent with t2h / t2f
- t2f, t2h and bridge are the **most reactive sites**

^aC. Díaz *et al.* Phys. Chem. Chem. Phys. **12**, 6499 (2010)

b J. Dai and J.C. Light, J. Chem. Phys. **107**, 1676 (1997)